

# USE OF NITROGEN BUDGETS AND N<sub>2</sub> FLUX MEASUREMENTS TO ESTIMATE THE ROLE OF DENITRIFICATION IN BROWNFIELD STORMWATER WETLANDS

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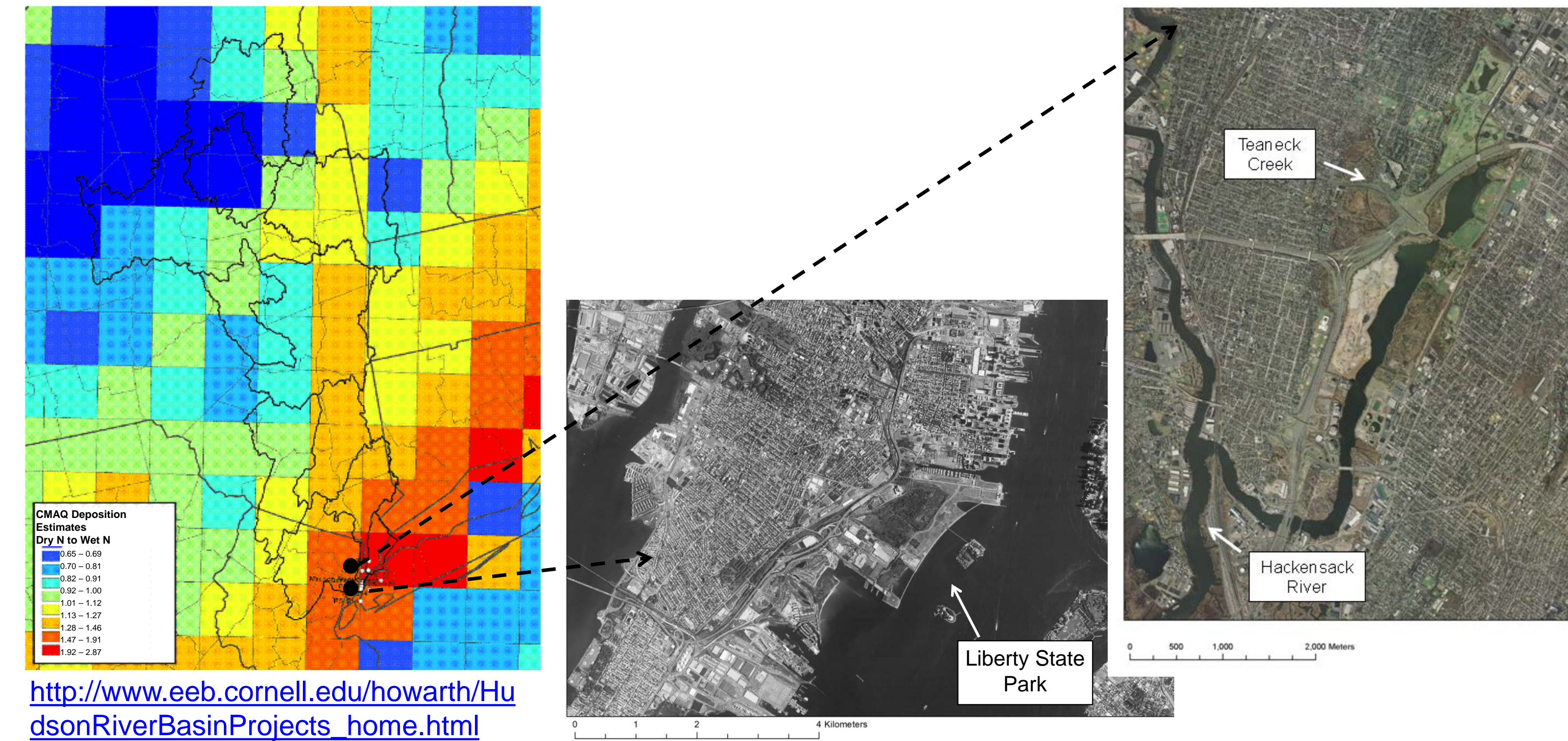
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## INTRODUCTION

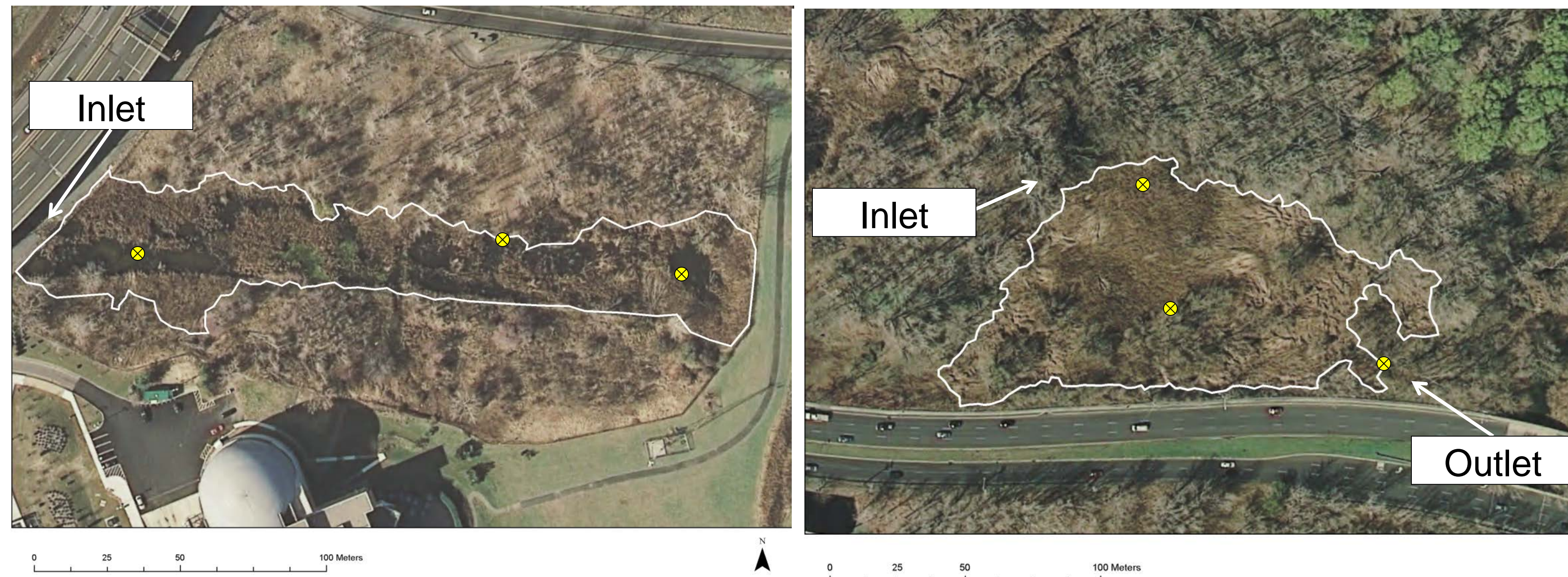
- Urban areas are net sources of excess inorganic nitrogen in waterways
- Wetlands provide desirable ecosystem services such as NO<sub>3</sub><sup>-</sup> removal via denitrification (microbial conversion of NO<sub>3</sub><sup>-</sup> to N<sub>2</sub>)
- Despite a broad understanding of environmental factors controlling rates of denitrification, few robust & predictive models have been constructed and validated for wetlands
- N<sub>2</sub> flux is difficult to measure, resulting in primarily measurements of potential or incomplete denitrification rates

## STUDY OBJECTIVES

- (1) Utilize **in situ** measurements of N<sub>2</sub> gas production and NO<sub>3</sub><sup>-</sup> loss in sediment profiles to calculate denitrification rates
- (2) Examine the **role and importance of denitrification** in inorganic nitrogen cycling and removal in urban brownfield wetlands

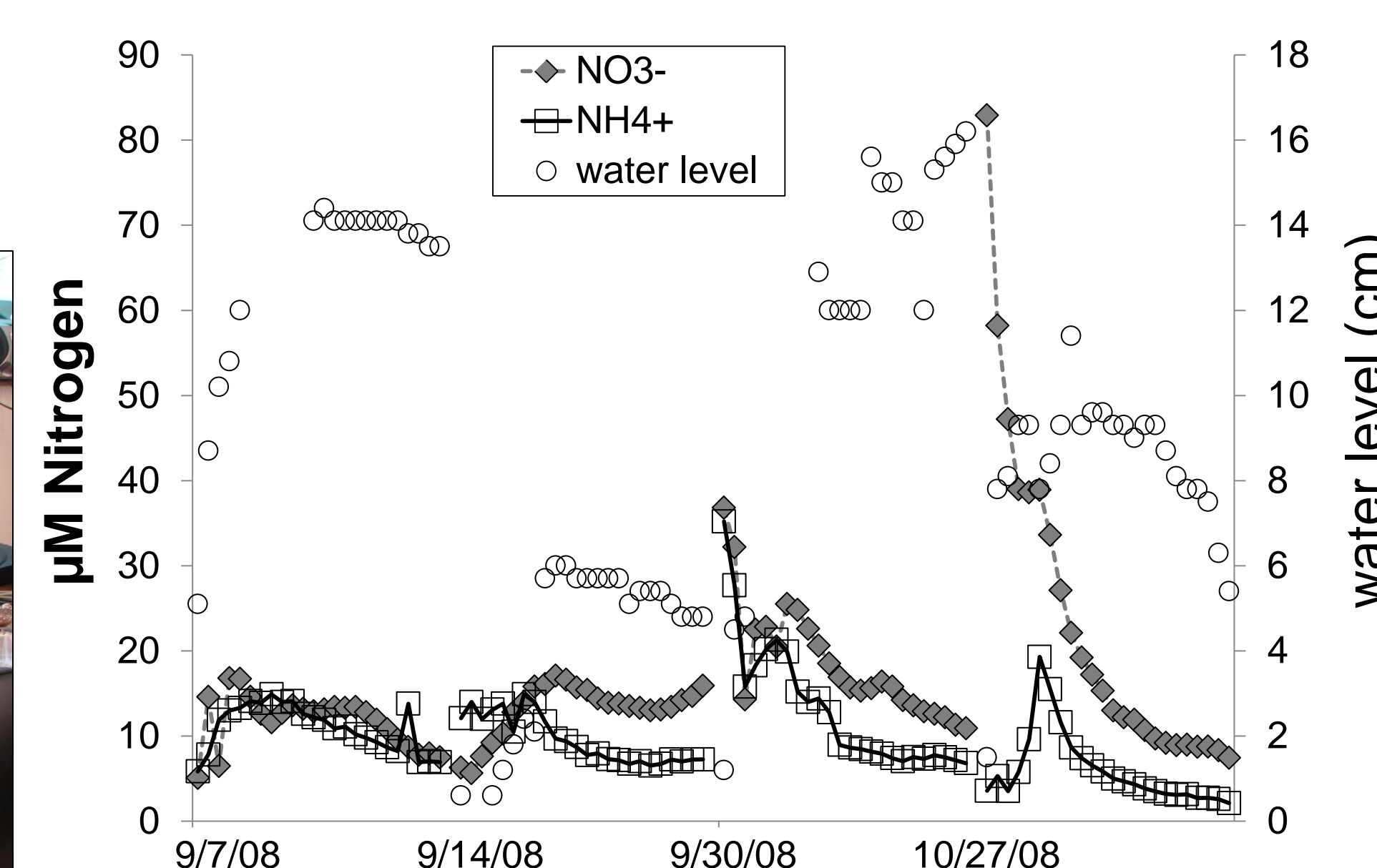


**INORGANIC N INPUTS IN THE NORTHEASTERN US ARE HIGH DUE TO URBAN STORMWATER AND ATMOSPHERIC DEPOSITION.** Rivers and estuaries in the region are subject to associated eutrophication problems; public parks are often an intermediary between urban stormwater from upland areas and adjacent waterbodies.

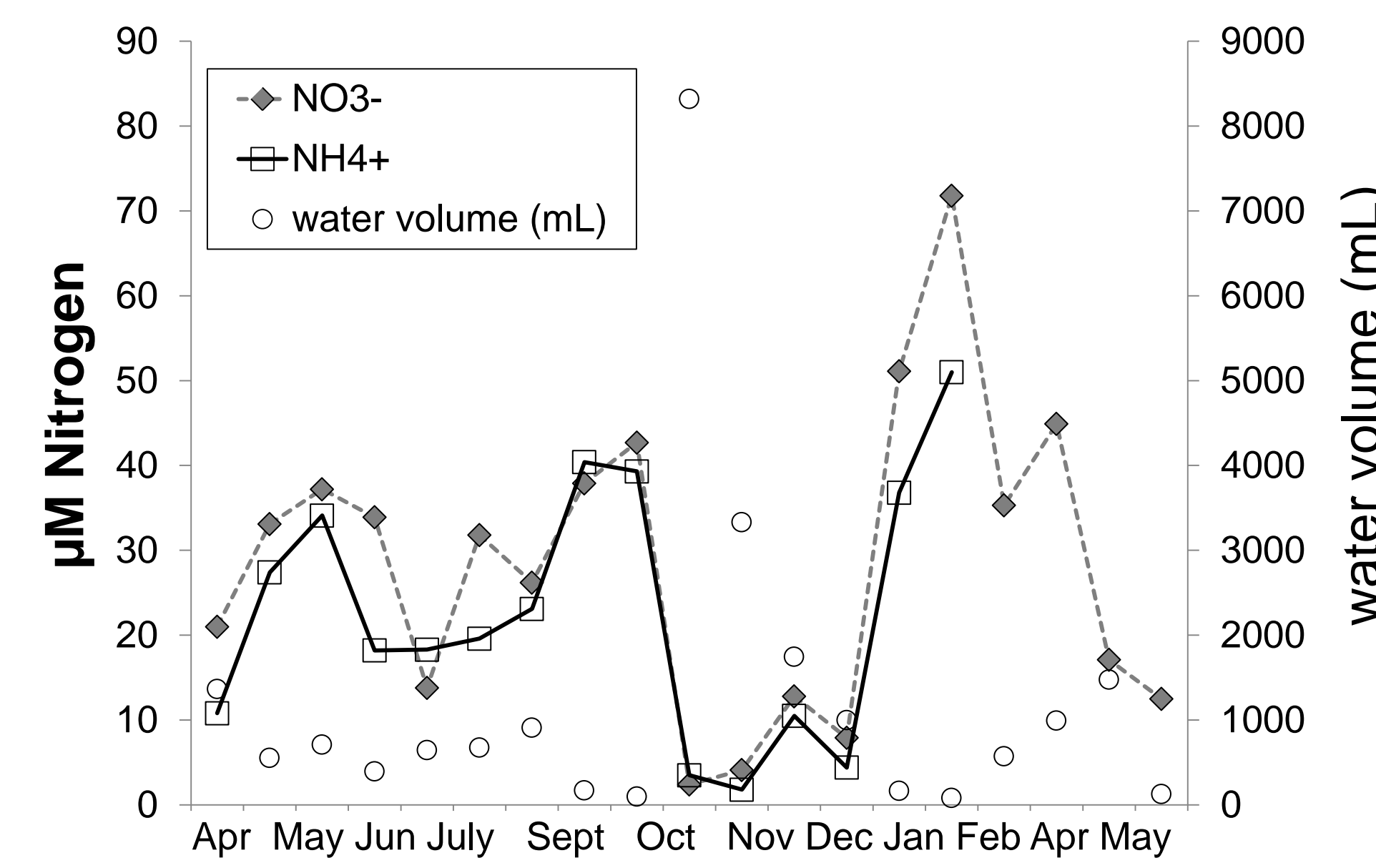


**LIBERTY STATE PARK AND TEANECK CREEK CONSERVANCY ARE URBAN BROWNFIELD SITES SUPPORTING SEMI-PERMANENTLY FLOODED WETLANDS.** White outlines delineate low-lying semi-permanently flooded areas. Flow in both wetlands moves west to east—surface water at the far left side of each photograph.

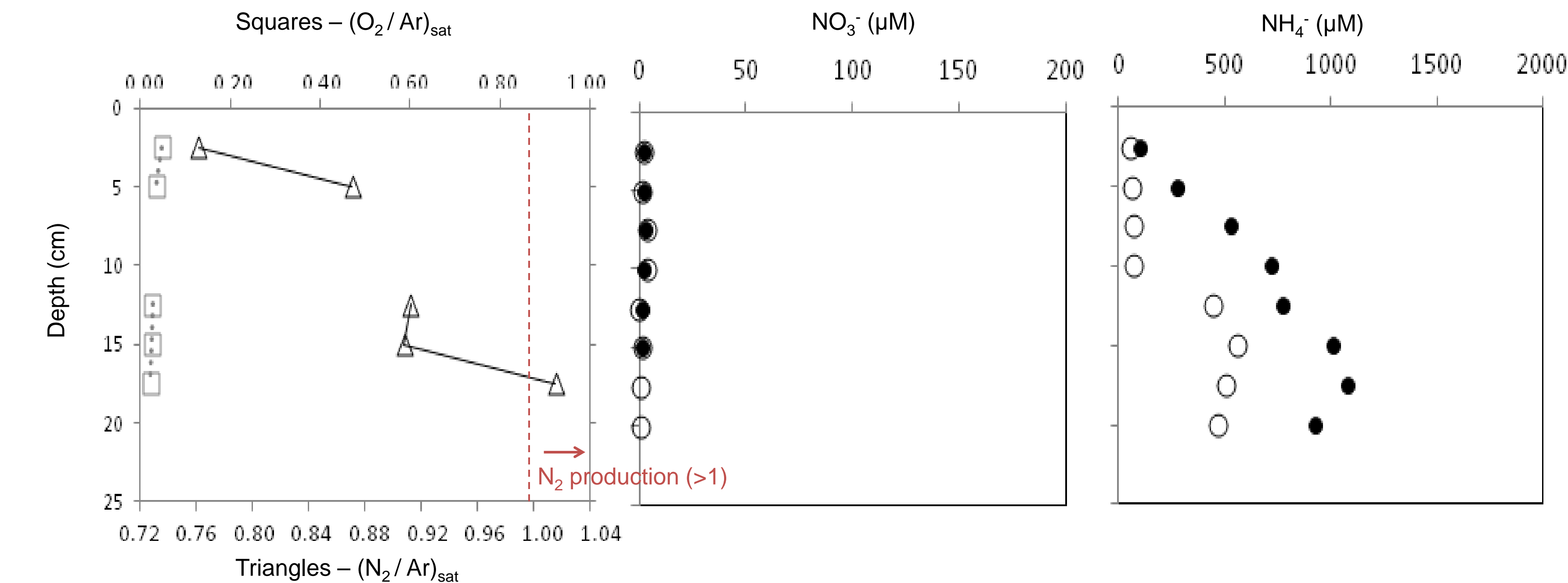
STORMWATER MONITORING AT LIBERTY, 2008



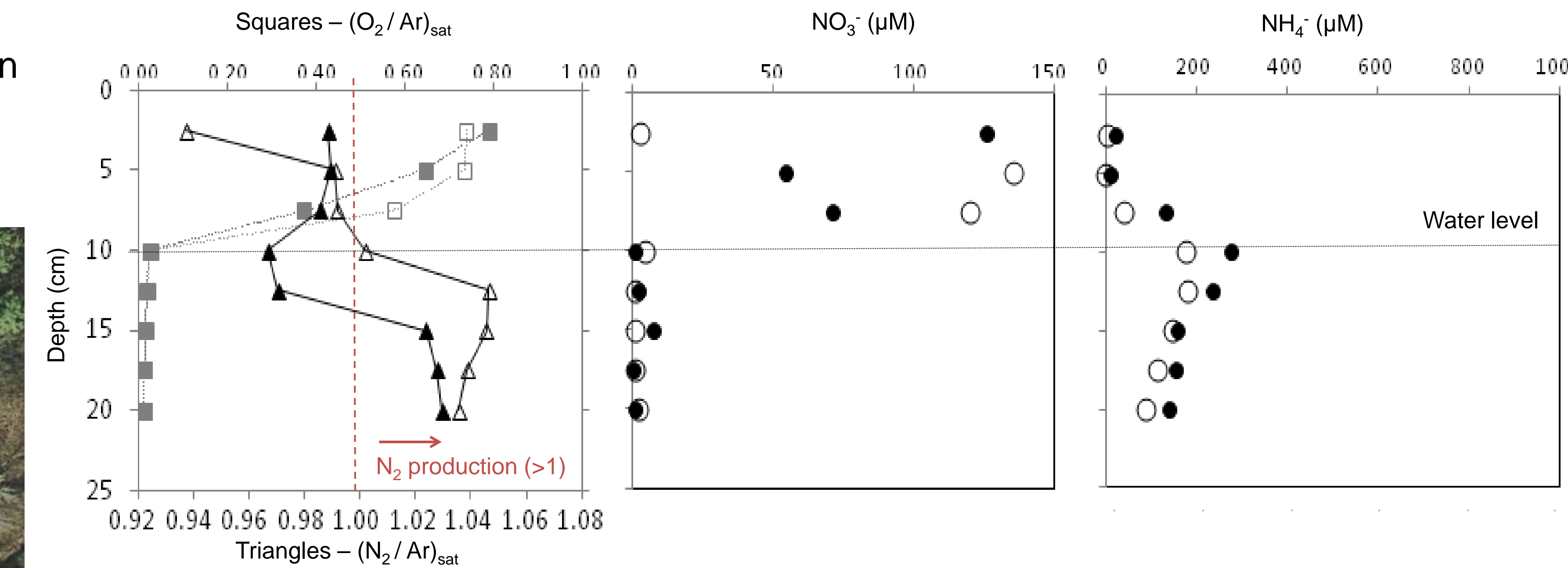
WET DEPOSITION MONITORING AT TEANECK, 2006



**NH<sub>4</sub><sup>+</sup> AND NO<sub>3</sub><sup>-</sup> LOADING IS HIGH DURING STORM EVENTS, AND EXHIBIT SIMILAR CONCENTRATIONS IN STORM AND RAINWATER ENTERING THE WETLANDS.**



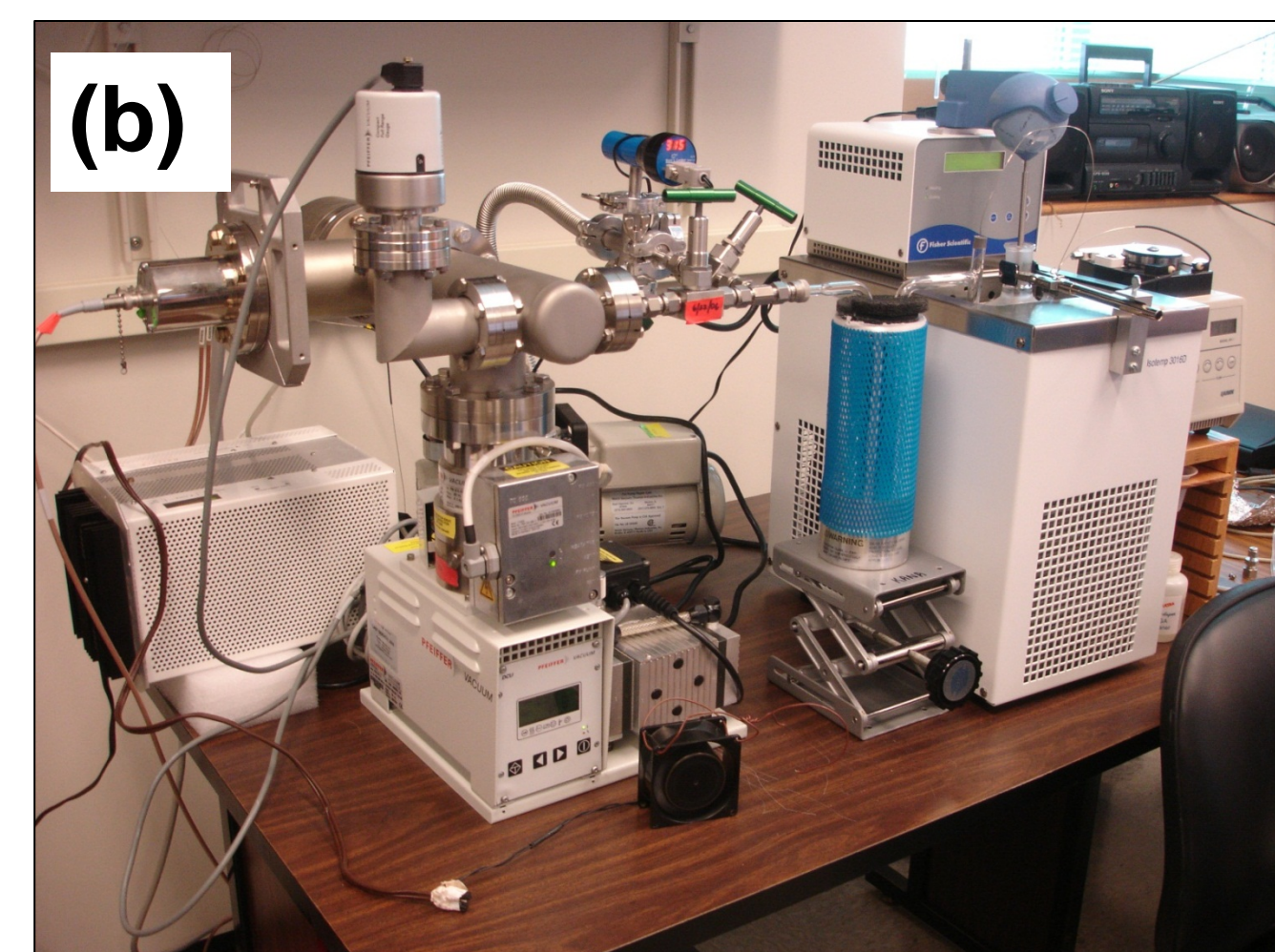
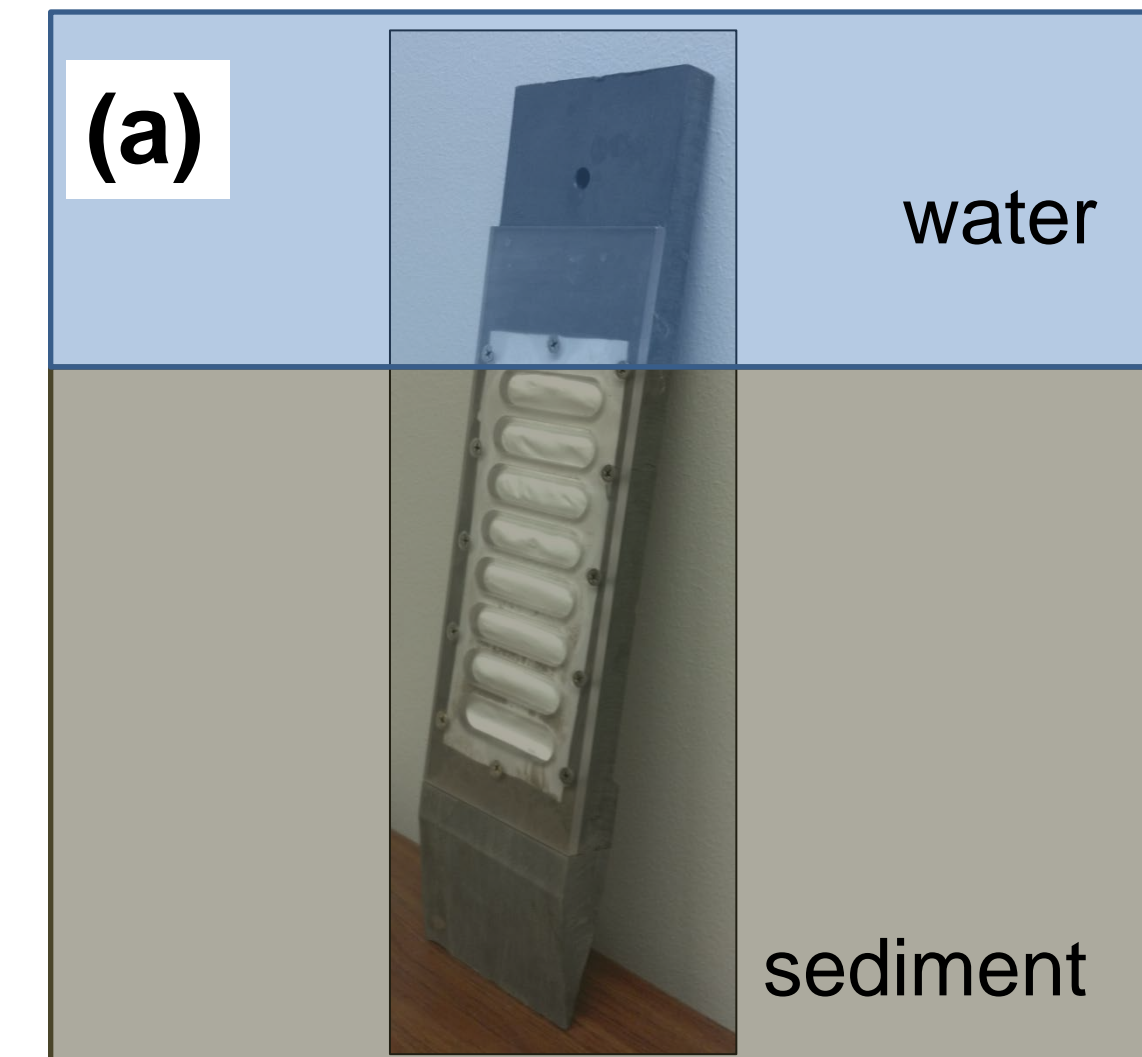
**FLOODED CONDITIONS RESULT IN LOW NO<sub>3</sub><sup>-</sup>, LOW N<sub>2</sub>, AND HIGH NH<sub>4</sub><sup>+</sup> PRODUCTION.** Peepers under flooded conditions demonstrated low dissolved O<sub>2</sub>, which appeared to promote NH<sub>4</sub><sup>+</sup> production via mineralization of plant material, but little denitrification due to low NO<sub>3</sub><sup>-</sup> availability (example from Liberty shown).



**DRY-DOWN CONDITIONS RESULT IN HIGH NO<sub>3</sub><sup>-</sup> PRODUCTION ABOVE THE SEDIMENT-WATER INTERFACE, AND HIGH N<sub>2</sub> PRODUCTION BELOW THE SEDIMENT-WATER INTERFACE.** Peepers under dry-down conditions demonstrated lower overall NH<sub>4</sub><sup>+</sup> production. N<sub>2</sub> production was much higher than under flooded conditions, and coincided with a drop in dissolved O<sub>2</sub> and NO<sub>3</sub><sup>-</sup> at the sediment-water interface (example from Teaneck shown).

## METHODS

- Denitrification measurements:
  - Peepers (a) to collect pore water in sediment (2009–2010)
  - Membrane Inlet Mass Spectrometry (MIMS) to analyze pore water for N<sub>2</sub>, Ar, and O<sub>2</sub> (b)
- Nitrogen loading to wetlands:
  - Atmospheric dry and wet deposition concentrations determined using literature values (Liberty) and a 2005–2006 field monitoring study (Teaneck)
  - Stormwater concentrations monitored at Liberty in 2008 and at Teaneck in 2005–2006
  - Volumes of stormwater entering and exiting wetlands estimated using SWMM and Mike SHE/Mike 11 models
  - Volumes of precipitation calculated from NOAA gages
- Calculations:
  - Saturation normalized N<sub>2</sub>/Ar ratio calculated using the following equation:  $(N_2/Ar)_{sat} = (N_2/Ar)_{molar\ ratio} / (N_2/Ar)_{saturation\ equilibrium\ ratio}$
  - Diffusive N<sub>2</sub>, NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup> fluxes in sediments calculated at the point of maximum slope of all dissolved constituents in the profile and where  $(N_2/Ar)_{sat} > 1.0$



	Dry Conditions	Wet Conditions
<b>Surface Water N Fluxes (TN)</b>		
Dry Deposition		
N-NO <sub>3</sub> <sup>-</sup>	3,090	3,090
N-NH <sub>4</sub> <sup>+</sup>	4,511	4,511
Wet Deposition		
N-NO <sub>3</sub> <sup>-</sup>	[Vol: 36.0 L] 2,592–34,380	[Vol: 57.3 L] 4,125–54,722
N-NH <sub>4</sub> <sup>+</sup>	[Vol: 35.2 L] 576–48,456	[Vol: 237.9 L] 917–77,125
Overland flow		
N-NO <sub>3</sub> <sup>-</sup>	[Vol: 35.2 L] 2,498–40,853	16,886–276,107
N-NH <sub>4</sub> <sup>+</sup>	[Vol: 35.2 L] 1,045–17,347	7,061–117,237
<b>Fluxes Between Water and Sediment</b>		
N-NO <sub>3</sub> <sup>-</sup> (OUT water pool, IN sediment pool)	3 640–1,900	590–630
Denitrification N-N <sub>2</sub> (IN atmosphere, OUT sediment pool)	5 1,000–2,580	620
N-NH <sub>4</sub> <sup>+</sup> (IN water, OUT sediment)	4 3,740–8,160	3,050–6,530

**NITROGEN BUDGETS REVEALED THAT WETLANDS ARE A SINK FOR NO<sub>3</sub><sup>-</sup> AND A SOURCE OF NH<sub>4</sub><sup>+</sup>.** Budget from Liberty shown here. Denitrification did not account for a large amount of nitrogen loss from the wetlands, likely because denitrification is limited by NO<sub>3</sub><sup>-</sup> availability

NH<sub>4</sub><sup>+</sup> produced in sediment dissolves into porewater and exits wetland via groundwater

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**REFERENCES.** Hartnett, H.E. and S.P. Seitzinger. 2003. High-resolution nitrogen gas profiles in sediment porewaters using a new membrane probe for membrane-inlet mass spectrometry. *Marine Chemistry* 83: 23-30; Gao, Y., Kennish, M. J., and A.M. Flynn. 2007. Atmospheric nitrogen deposition to the New Jersey coastal waters and its implications. *Ecological Applications* 17: S31-S41; Song, F. and Y. Gao. 2009. Chemical characteristics of precipitation at metropolitan Newark in the US East Coast. *Atmospheric Environment* 43: 4903-4913.